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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.003 Å R factor = 0.041 wR factor = 0.107 Data-to-parameter ratio = 14.0

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

1'-Methyl-5-diphenyl-4'-p-tolyl-5,6,7,8,9,10-hexahydro-1,3-cycloheptapyrimidino[2,3-b]thiazole-2-spiro-3'-pyrrolidine-2'-spiro-3"-1H-indole-2",3(2H,3"H)-dione

The pyrrolidine ring in the title compound, $C_{35}H_{34}N_4O_2S$, adopts an envelope conformation, with the N atom deviating by 0.586 (1) Å from the plane of the other atoms. The molecule is stabilized by weak $C-H\cdots O$ interactions and the crystal packing is stabilized by $N-H\cdots N$ intermolecular interactions, generating a dimer with an $R_2^2(16)$ motif.

Comment

Heterocyclic compounds, especially five- and six-membered rings, have occupied an important place among organic compounds for their great pharmaceutical importance, especially pyrrolidine and thiazolidine derivatives. Synthetic spiropyrrolidine derivatives show activity against the aldose reductase enzyme which controls influenza (Stylianakis *et al.*, 2003). Indole and its derivatives represent one of the most active classes of compounds, possessing a wide range of biological activities (Hiremath *et al.*, 1988). Owing to the ease of substitution and modifications at several positions, many derivatives of pyrrolidine have been synthesized with different properties (Baldwin *et al.*, 1994). In view of the importance of these compounds, we have undertaken the structure determination of the title compound, (I) (Fig. 1).



The molecular geometry of (I) is comparable with that of a related structure reported earlier (Gayathri *et al.*, 2005). The sums of the bond angles around N1 (337.2°) and N3 (359.9°) indicate sp^3 and sp^2 hybridization, respectively.

The methyl atom C5 lies 0.545 (2) Å below the plane of atoms C1–C4, and atom C35 lies 0.087 (3) Å above the plane of the benzene ring C29–C34. The five- (C1/C6/N2/C7/C12) and six- (C7-C12) membered rings in the indane group are

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Figure 1

The molecular structure of the title compound, showing 30% probability displacement ellipsoids.





The packing of (I), viewed approximately down the *a* axis. Hydrogen bonds are shown as dashed lines. H atoms not involved in hydrogen bonding have been omitted.

planar, with a dihedral angle of $6.3 (1)^{\circ}$ between these rings. Atom O1 deviates by 0.105 (1) Å from the plane of the fivemembered ring in the indane group. The six-membered ring N3/C14/C21/C27/N4/C28 is slightly non-planar, with atom C14 deviating by 0.159 (2) Å from the plane of the other atoms, because of the phenyl (C15-C20) substituent at atom C14. The dihedral angle between the two benzene rings (C15-C20 and C29-C34) is 21.9 (1)°.

The pyrrolidine ring adopts an envelope conformation, with atom N1 deviating by 0.586 (1) Å from the plane of the other atoms. The puckering parameters (Cremer & Pople, 1975) and the smallest displacement asymmetry parameter (Nardelli, 1983) for the pyrrolidine ring are $q_2 = 0.406$ (2) Å, $\varphi =$ 353.4 (3)° and $\Delta_s(N_1) = 6.7$ (2).

The molecule is stabilized by weak C-H···O intramolecular interactions and the crystal packing is stabilized by N-H···N intermolecular interactions, generating a centrosymmetric dimer of $R_2^2(16)$ motif (Bernstein *et al.*, 1995) (Table 1 and Fig. 2).

Experimental

A mixture of isatin (1.2 mmol), sarcosine (1.2 mmol) and 5-phenyl-2-(p-methyl)phenylmethylene-5,6,7,8,9,10-hexahydrocyclohepta[d]thiazolo[3,2-a]pyrimidin-3(2H)-one (1 mmol) in methanol-dioxane (1:1, 20 ml) was refluxed until the disappearance of the starting materials (5 h) as shown by thin-layer chromatography analysis. The reaction mixture was then concentrated in vacuo and extracted with water (50 ml) and dichloromethane (50 ml). The organic layer was washed with brine, dried with anhydrous sodium sulfate and concentrated in vacuo. The residue was purified by column chromatography (silica gel, 100-200 mesh) eluted with a hexane-ethyl acetate (8:2) mixture to give the title compound, which was recrystallized from methanol by slow evaporation.

Crystal data

γ

$C_{35}H_{34}N_4O_2S$	$V = 1512.67 (19) \text{ Å}^3$
$M_r = 574.72$	Z = 2
Triclinic, P1	$D_x = 1.262 \text{ Mg m}^{-3}$
a = 11.3427 (8) Å	Mo $K\alpha$ radiation
b = 12.0148 (9) Å	$\mu = 0.14 \text{ mm}^{-1}$
c = 13.0102 (10) Å	T = 293 (2) K
$\alpha = 95.604 \ (1)^{\circ}$	Block, colourless
$\beta = 110.887 \ (1)^{\circ}$	$0.24 \times 0.21 \times 0.20 \text{ mm}$
$\gamma = 109.567 \ (1)^{\circ}$	

Data collection

Bruker SMART APEX CCD areadetector diffractometer ω scans Absorption correction: none 14702 measured reflections

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0515P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.041$	+ 0.4411P]
$wR(F^2) = 0.107$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.04	$(\Delta/\sigma)_{\rm max} < 0.001$
5317 reflections	$\Delta \rho_{\rm max} = 0.24 \text{ e } \text{\AA}^{-3}$
379 parameters	$\Delta \rho_{\rm min} = -0.17 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	

Table 1 Hydrogen-bond geometry (Å, °).

D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
0.98	2.45	2.944 (2)	111
0.97	2.50	3.070 (2)	118
0.93	2.46	3.045 (3)	121
0.86	2.13	2.988 (2)	177
	<i>D</i> -H 0.98 0.97 0.93 0.86	$D-H$ $H \cdots A$ 0.98 2.45 0.97 2.50 0.93 2.46 0.86 2.13	$D-H$ $H \cdots A$ $D \cdots A$ 0.982.452.944 (2)0.972.503.070 (2)0.932.463.045 (3)0.862.132.988 (2)

Symmetry code: (i) -x + 1, -y, -z + 1.

All H atoms were treated using a riding model, with C-H = 0.93for aromatic H, 0.98 for methine H, 0.97 for methylene H and 0.96 Å for methyl H, and N-H = 0.86 Å, and with $U_{iso}(H) = 1.2U_{eq}(C,N)$, or $1.5U_{eq}(C)$ for methyl groups.

5317 independent reflections 4625 reflections with $I > 2\sigma(I)$

 $R_{\rm int} = 0.021$

 $\theta_{\rm max} = 25.0^{\circ}$

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97* and *PARST* (Nardelli, 1995).

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